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Chapter 12 Stoichiometry 293 Name ____ Date ____ Class ____ LIMITING REAGENT AND PERCENT YIELD 12.3

05 CTR ch12 7/9/04 3:34 PM Page 289 THE ARITHMETIC OF ...

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Chapter 12

Chapter 12 Stoichiometry127 SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353–358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

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Chapter 3 Chemical Reactions and Reaction Stoichiometry

Chapter 3 Stoichiometry: Calculations with Chemical Formulas and Equations 3.1 Multiple -Choice Questions 1) When the following equation is balanced, the coefficients are _____. $C_8H_{18} + O_2 \rightarrow CO_2 + H_2O$ A) 2, 3, 4, 4 B) 1, 4, 8, 9 C) 2, 12, 8, 9 D) 4, 4, 32, 36 E) 2, 25, 16, 18 Answer: E Diff: 2 Page Ref: Sec. 3.1

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12.3 Limiting Reagent and Percent Yield > ... percent yield:

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